

Dip Tank Cleaner Solution Disposal Instructions

Safety First:	Wear protective gear, including rubber gloves, apron, and goggles, when working with Dip Tank Cleaner. Handle vinegar with care to avoid any potential skin or eye irritation.
Safe Disposal:	Check local regulations and guidelines regarding the disposal of chemical solutions. Neutralize the solution completely and allow it to cool before pouring it down the drain.
	Remember that safety precautions are crucial when handling Dip Tank Cleaner, and proper disposal practices are essential to protect the environment and ensure safety.
pH Test Strips:	Obtain 1-14 pH test strips from HIX Corporation (Part #50481) or from a local swimming pool (chemical) supply store.
Vinegar Selection:	Most types of vinegar, including white vinegar, apple cider vinegar, and wine vinegar, can effectively neutralize Dip Tank Cleaner due to their acidic nature.
	When vinegar and Dip Tank Cleaner solution come into contact, a neutralization reaction occurs. This reaction helps to reduce the corrosive properties of Dip Tank Cleaner solution, making it less harmful to surfaces, fabrics, and skin.
Solution Prep:	Measure the amount (gallons) of Dip Tank Cleaner solution that needs neutralization.
	Dilute the Dip Tank Cleaner solution if it's highly concentrated. Add water to reduce its strength.
Check pH Levels:	Use pH indicator paper or a pH meter to check the pH of the solution.
	The goal is to achieve a neutral pH (around 7 is ideal, a range of 6.5 to 8.5 is acceptable) before pouring the solution down the drain.
Mix Solution:	Mix Vinegar into the Dip Tank Cleaner solution (26104 QNC Dip Tank Cleaner):
	Gradually and slowly add vinegar to the Dip Tank Cleaner solution.
	Aim for a 1:1 ratio of vinegar to Dip Tank Cleaner. However, it's essential to conduct a small test in a controlled environment (e.g. a gallon test sample) to ensure the desired result.
Reaction:	As you add vinegar, stir slowly, and observe the solution, checking the pH frequently.
	The neutralization process will result in the formation of water and sodium acetate, reducing the corrosive properties of the the Dip Tank Cleaner.





